

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

4-[5-(4-Fluorophenyl)-1*H*-imidazol-4-yl]-pyridinePierre Koch,^a Dieter Schollmeyer^b and Stefan Laufer^{a*}

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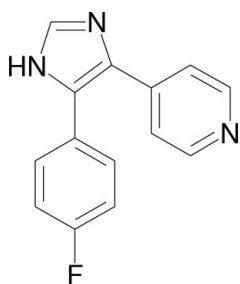
Received 26 January 2009; accepted 17 February 2009

Key indicators: single-crystal X-ray study; $T = 193$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.076; wR factor = 0.201; data-to-parameter ratio = 13.0.

In the title compound, $\text{C}_{14}\text{H}_{10}\text{FN}_3$, the imidazole ring makes dihedral angles of 28.2 (1) and 36.60 (9)° with the pyridine ring and the 4-fluorophenyl ring, respectively. The pyridine ring forms a dihedral angle of 44.68 (9)° with the 4-fluorophenyl ring. Intermolecular $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonds are observed in the crystal structure.

Related literature

For the biological activity of the title compound, see: Liverton *et al.* (1999). For applications of functionalized 5(4)-(4-fluorophenyl)-4(5)-(pyridin-4-yl)imidazoles, see: Koch *et al.* (2008), Peifer *et al.* (2006).



Experimental

Crystal data

 $\text{C}_{14}\text{H}_{10}\text{FN}_3$ $M_r = 239.25$ Orthorhombic, $Pbca$

$a = 9.217$ (2) Å
 $b = 8.1064$ (5) Å
 $c = 30.665$ (5) Å
 $V = 2291.1$ (6) Å³

 $Z = 8$

Cu $K\alpha$ radiation
 $\mu = 0.80$ mm⁻¹
 $T = 193$ K
 $0.54 \times 0.20 \times 0.13$ mm

Data collection

Enraf-Nonius CAD-4 diffractometer
 Absorption correction: none
 2121 measured reflections
 2121 independent reflections

1707 reflections with $I > 2\sigma(I)$
 3 standard reflections
 frequency: 60 min
 intensity decay: 2%

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.076$
 $wR(F^2) = 0.201$
 $S = 1.09$
 2121 reflections

163 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.58$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.54$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1}\cdots\text{N15}^i$	0.89	1.94	2.815 (3)	164

Symmetry code: (i) $x - 1, y, z$.

Data collection: *CAD-4 Software* (Enraf-Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *CORINC* (Dräger & Gattow, 1971); program(s) used to solve structure: *SIR97* (Altomare *et al.*, 1999); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *PLATON*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: IM2099).

References

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supplementary materials

Acta Cryst. (2009). E65, o573 [doi:10.1107/S1600536809005650]

4-[5-(4-Fluorophenyl)-1*H*-imidazol-4-yl]pyridine

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Comment

5(4)-(4-Fluorophenyl)-4(5)-(pyridin-4-yl)imidazole derivatives with various substitution patterns have been considered to be potential p38 MAP kinase inhibitors (Liverton *et al.* 1999, Koch *et al.* 2008, Peifer *et al.* 2006).

The molecular structure of compound **I** is shown in Figure 1. The imidazole ring realises dihedral angles of 28.2 (1) $^{\circ}$ and 36.60 (9) $^{\circ}$ with the pyridine ring and the 4-fluorophenyl ring, respectively. The pyridine ring encloses a dihedral angle of 44.68 (9) $^{\circ}$ with the 4-fluorophenyl ring.

The crystal packing (Figure 2) shows N1—H1 of the imidazole ring to form an intermolecular N—H \cdots N hydrogen bond towards pyridine (N15) resulting in a infinite chain parallel to the *a* axis. The hydrogen bond measures 1.94 Å.

Experimental

1-(4-Fluorophenyl)-2-(pyridin-4-yl)ethane-1,2-dione (46 mg, 0.2 mmol), formaldehyde (15 μ L, 0.2 mmol, 37% aq. solution), ammonium acetate (154 mg, 2.0 mmol) and 1 ml glacial acetic acid were combined in a reaction vial. The reaction vessel was heated in a CEM microwave reactor for 5 min at 453 K (initial power 200 W), after which a stream of compressed air cooled the reaction vessel. The reaction mixture was added dropwise to a concentrated NH₄OH solution at 0 $^{\circ}$ C. The formed colorless precipitate was collected by filtration, washed with water and dried (yield: 43 mg, 90%). Crystals of compound **I** suitable for X-ray diffraction were obtained by slow evaporation at 298 K of a solution of n-hexane - diethyl ether (3:2).

Refinement

Hydrogen atoms attached to carbons were placed at calculated positions with C—H = 0.95 Å (aromatic C-atoms). The position of H1 was determined from the difference Fourier map. All H atoms were refined in the riding-model approximation with isotropic displacement parameters (set at 1.2 times of the U_{eq} of the parent atom).

Figures

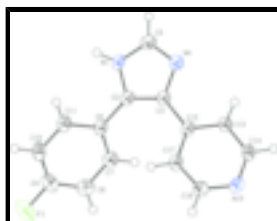


Fig. 1. View of compound **I**. Displacement ellipsoids are drawn at the 50% probability level. H atoms are depicted as circles of arbitrary size.

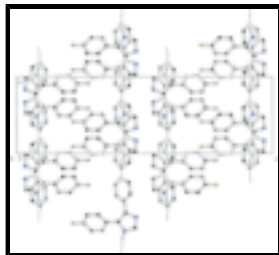


Fig. 2. Part of the crystal packing of compound I. The hydrogen bonds are represented by dashed lines. View along *b* axis.

4-[5-(4-Fluorophenyl)-1*H*-imidazol-4-yl]pyridine

Crystal data

$C_{14}H_{10}FN_3$

$M_r = 239.25$

Orthorhombic, *Pbca*

Hall symbol: -P 2ac 2ab

$a = 9.217 (2) \text{ \AA}$

$b = 8.1064 (5) \text{ \AA}$

$c = 30.665 (5) \text{ \AA}$

$V = 2291.1 (6) \text{ \AA}^3$

$Z = 8$

$F_{000} = 992$

$D_x = 1.387 \text{ Mg m}^{-3}$

Melting point: 285.5 K

Cu $K\alpha$ radiation

$\lambda = 1.54178 \text{ \AA}$

Cell parameters from 25 reflections

$\theta = 31\text{--}53^\circ$

$\mu = 0.80 \text{ mm}^{-1}$

$T = 193 \text{ K}$

Needle, colourless

$0.54 \times 0.20 \times 0.13 \text{ mm}$

Data collection

Enraf-Nonius CAD-4
diffractometer

Monochromator: graphite

$T = 193 \text{ K}$

$\omega/2\theta$ scans

Absorption correction: none

2121 measured reflections

2121 independent reflections

1707 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.0000$

$\theta_{\text{max}} = 69.6^\circ$

$\theta_{\text{min}} = 2.9^\circ$

$h = 0 \rightarrow 11$

$k = 0 \rightarrow 9$

$l = -36 \rightarrow 0$

3 standard reflections

every 60 min

intensity decay: 2%

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.076$

$wR(F^2) = 0.201$

$S = 1.09$

2121 reflections

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.1422P)^2 + 0.0554P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} < 0.001$

$\Delta\rho_{\text{max}} = 0.58 \text{ e \AA}^{-3}$

163 parameters

$$\Delta\rho_{\min} = -0.54 \text{ e } \text{\AA}^{-3}$$

Primary atom site location: structure-invariant direct methods

Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
F1	0.0873 (2)	0.0708 (3)	0.21998 (6)	0.0596 (6)
N1	-0.0150 (2)	0.2988 (2)	0.42177 (7)	0.0243 (5)
H1	-0.1070	0.2883	0.4130	0.029*
C2	0.1085 (2)	0.2755 (3)	0.39720 (8)	0.0224 (5)
C3	0.2212 (3)	0.3123 (3)	0.42585 (8)	0.0227 (5)
N4	0.1672 (2)	0.3595 (3)	0.46674 (7)	0.0285 (5)
C5	0.0257 (3)	0.3479 (3)	0.46279 (9)	0.0279 (6)
H5	-0.0408	0.3708	0.4857	0.033*
C6	0.1015 (2)	0.2250 (3)	0.35030 (8)	0.0228 (5)
C7	0.1984 (3)	0.2872 (3)	0.31898 (9)	0.0286 (6)
H7	0.2682	0.3669	0.3277	0.034*
C8	0.1955 (3)	0.2353 (3)	0.27519 (9)	0.0338 (6)
H8	0.2639	0.2764	0.2547	0.041*
C9	0.0907 (3)	0.1233 (3)	0.26276 (9)	0.0365 (7)
C10	-0.0096 (3)	0.0625 (3)	0.29215 (9)	0.0353 (6)
H10	-0.0819	-0.0131	0.2828	0.042*
C11	-0.0043 (3)	0.1128 (3)	0.33599 (8)	0.0274 (6)
H11	-0.0731	0.0705	0.3562	0.033*
C12	0.3785 (3)	0.3036 (3)	0.41935 (8)	0.0216 (5)
C13	0.4700 (3)	0.4057 (3)	0.44398 (8)	0.0246 (5)
H13	0.4299	0.4814	0.4643	0.030*
C14	0.6182 (3)	0.3961 (3)	0.43855 (8)	0.0282 (6)
H14	0.6783	0.4650	0.4559	0.034*
N15	0.6824 (2)	0.2927 (3)	0.40951 (7)	0.0287 (5)
C16	0.5937 (3)	0.1937 (3)	0.38606 (9)	0.0290 (6)
H16	0.6367	0.1193	0.3659	0.035*
C17	0.4456 (3)	0.1942 (3)	0.38966 (8)	0.0264 (6)
H17	0.3885	0.1218	0.3724	0.032*

supplementary materials

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
F1	0.0846 (15)	0.0588 (13)	0.0354 (10)	-0.0016 (11)	-0.0071 (9)	-0.0100 (9)
N1	0.0210 (10)	0.0106 (9)	0.0414 (12)	-0.0012 (7)	0.0000 (8)	-0.0015 (8)
C2	0.0225 (11)	0.0050 (10)	0.0398 (14)	0.0003 (7)	0.0019 (9)	0.0031 (9)
C3	0.0277 (13)	0.0048 (10)	0.0355 (12)	-0.0005 (8)	0.0005 (9)	0.0007 (8)
N4	0.0293 (11)	0.0182 (10)	0.0379 (12)	-0.0021 (8)	0.0020 (9)	-0.0029 (8)
C5	0.0273 (12)	0.0166 (11)	0.0399 (14)	-0.0008 (9)	0.0058 (10)	-0.0027 (10)
C6	0.0234 (11)	0.0078 (10)	0.0371 (13)	0.0037 (8)	-0.0022 (9)	0.0011 (9)
C7	0.0308 (12)	0.0139 (11)	0.0410 (15)	0.0016 (9)	-0.0012 (10)	0.0027 (10)
C8	0.0379 (14)	0.0257 (13)	0.0377 (15)	0.0053 (10)	0.0035 (11)	0.0069 (11)
C9	0.0499 (17)	0.0278 (14)	0.0319 (14)	0.0083 (11)	-0.0085 (12)	-0.0012 (11)
C10	0.0383 (14)	0.0215 (12)	0.0460 (16)	-0.0033 (11)	-0.0113 (12)	-0.0041 (11)
C11	0.0277 (12)	0.0139 (11)	0.0407 (14)	-0.0017 (9)	-0.0044 (10)	0.0020 (9)
C12	0.0239 (12)	0.0070 (10)	0.0337 (13)	-0.0007 (8)	-0.0014 (9)	0.0044 (8)
C13	0.0282 (12)	0.0151 (11)	0.0306 (12)	-0.0020 (9)	-0.0004 (9)	-0.0006 (9)
C14	0.0294 (13)	0.0194 (12)	0.0357 (13)	-0.0036 (9)	-0.0039 (10)	-0.0016 (10)
N15	0.0231 (10)	0.0223 (11)	0.0408 (12)	0.0008 (8)	-0.0016 (9)	0.0022 (9)
C16	0.0286 (13)	0.0156 (12)	0.0429 (15)	0.0050 (9)	-0.0006 (10)	-0.0024 (10)
C17	0.0273 (12)	0.0091 (10)	0.0427 (15)	-0.0001 (9)	-0.0044 (10)	-0.0017 (9)

Geometric parameters (\AA , $^\circ$)

F1—C9	1.380 (3)	C8—H8	0.9500
N1—C5	1.372 (3)	C9—C10	1.382 (4)
N1—C2	1.378 (3)	C10—C11	1.406 (4)
N1—H1	0.8936	C10—H10	0.9500
C2—C3	1.393 (3)	C11—H11	0.9500
C2—C6	1.496 (3)	C12—C13	1.402 (3)
C3—N4	1.402 (3)	C12—C17	1.414 (3)
C3—C12	1.465 (3)	C13—C14	1.379 (3)
N4—C5	1.313 (3)	C13—H13	0.9500
C5—H5	0.9500	C14—N15	1.359 (3)
C6—C11	1.403 (3)	C14—H14	0.9500
C6—C7	1.405 (3)	N15—C16	1.353 (3)
C7—C8	1.407 (4)	C16—C17	1.370 (3)
C7—H7	0.9500	C16—H16	0.9500
C8—C9	1.379 (4)	C17—H17	0.9500
C5—N1—C2	108.4 (2)	F1—C9—C10	119.7 (3)
C5—N1—H1	124.3	C9—C10—C11	119.8 (2)
C2—N1—H1	127.3	C9—C10—H10	120.1
N1—C2—C3	104.0 (2)	C11—C10—H10	120.1
N1—C2—C6	121.9 (2)	C6—C11—C10	120.7 (2)
C3—C2—C6	134.2 (2)	C6—C11—H11	119.6
C2—C3—N4	111.0 (2)	C10—C11—H11	119.6
C2—C3—C12	129.9 (2)	C13—C12—C17	117.0 (2)

N4—C3—C12	119.0 (2)	C13—C12—C3	119.6 (2)
C5—N4—C3	104.5 (2)	C17—C12—C3	123.4 (2)
N4—C5—N1	112.1 (2)	C14—C13—C12	119.8 (2)
N4—C5—H5	123.9	C14—C13—H13	120.1
N1—C5—H5	123.9	C12—C13—H13	120.1
C11—C6—C7	117.4 (2)	N15—C14—C13	123.1 (2)
C11—C6—C2	120.5 (2)	N15—C14—H14	118.5
C7—C6—C2	122.1 (2)	C13—C14—H14	118.5
C6—C7—C8	122.2 (2)	C16—N15—C14	116.8 (2)
C6—C7—H7	118.9	N15—C16—C17	123.9 (2)
C8—C7—H7	118.9	N15—C16—H16	118.0
C9—C8—C7	118.3 (3)	C17—C16—H16	118.0
C9—C8—H8	120.9	C16—C17—C12	119.3 (2)
C7—C8—H8	120.9	C16—C17—H17	120.3
C8—C9—F1	118.8 (3)	C12—C17—H17	120.3
C8—C9—C10	121.5 (3)		
C5—N1—C2—C3	0.3 (2)	C7—C8—C9—C10	0.0 (4)
C5—N1—C2—C6	-178.6 (2)	C8—C9—C10—C11	-1.1 (4)
N1—C2—C3—N4	-0.9 (2)	F1—C9—C10—C11	178.5 (2)
C6—C2—C3—N4	177.8 (2)	C7—C6—C11—C10	1.6 (3)
N1—C2—C3—C12	177.2 (2)	C2—C6—C11—C10	-178.8 (2)
C6—C2—C3—C12	-4.1 (4)	C9—C10—C11—C6	0.2 (4)
C2—C3—N4—C5	1.2 (2)	C2—C3—C12—C13	153.7 (2)
C12—C3—N4—C5	-177.14 (19)	N4—C3—C12—C13	-28.4 (3)
C3—N4—C5—N1	-1.0 (3)	C2—C3—C12—C17	-27.5 (4)
C2—N1—C5—N4	0.4 (3)	N4—C3—C12—C17	150.5 (2)
N1—C2—C6—C11	-37.4 (3)	C17—C12—C13—C14	0.0 (3)
C3—C2—C6—C11	144.1 (2)	C3—C12—C13—C14	178.9 (2)
N1—C2—C6—C7	142.3 (2)	C12—C13—C14—N15	1.1 (4)
C3—C2—C6—C7	-36.3 (4)	C13—C14—N15—C16	-1.5 (4)
C11—C6—C7—C8	-2.7 (3)	C14—N15—C16—C17	0.9 (4)
C2—C6—C7—C8	177.6 (2)	N15—C16—C17—C12	0.1 (4)
C6—C7—C8—C9	2.0 (4)	C13—C12—C17—C16	-0.5 (3)
C7—C8—C9—F1	-179.6 (2)	C3—C12—C17—C16	-179.4 (2)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1 \cdots N15 ⁱ	0.89	1.94	2.815 (3)	164

Symmetry codes: (i) $x-1, y, z$.

Fig. 1

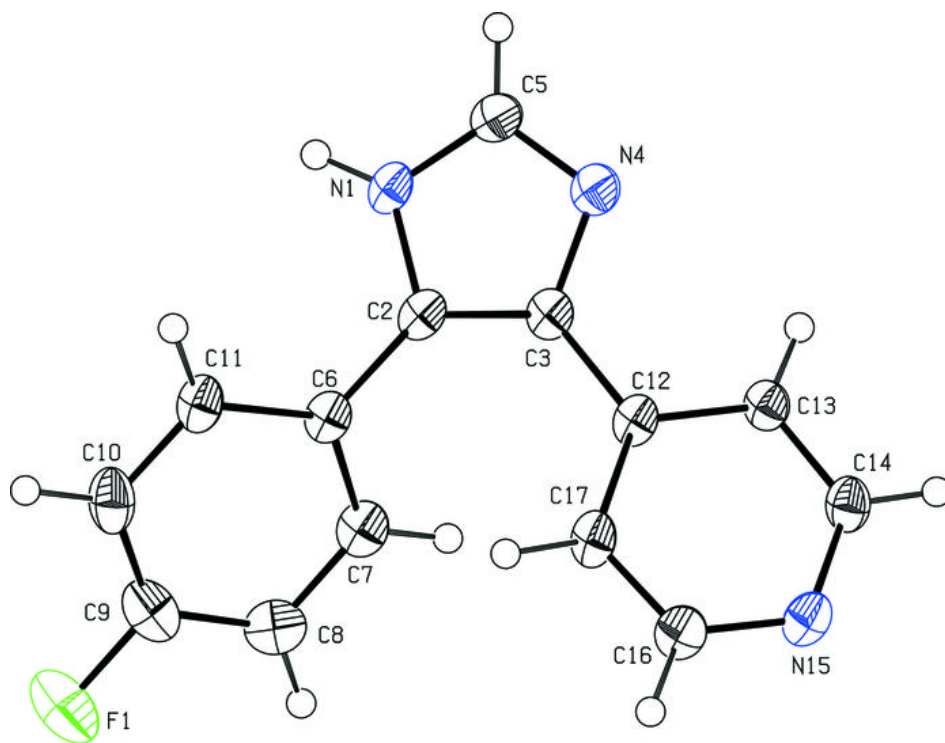


Fig. 2

